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Part A Standardization of KMnO_4 Solution 1) Make your Diluted KMnO_4 solution in the 500 mL Florence flask a) Add 25 mL of the concentrated KMnO_4 solution to the Florence flask b) Add 375 mL of H_2O to the Florence flask c) Mix well 2) Set up one buret and fill with the Diluted KMnO_4 3) To an Erlenmeyer flask add: a) ~0.70 g of $\text{Fe}(\text{NH}_4)_2(\text{SO}_4)_2 \cdot 6\text{H}_2\text{O}$ b) 30 mL of H_2O c) 10 mL of 3M H_2SO_4 4) Using the KMnO_4 buret Diluted a) Record the initial volume. KMnO_4 ...

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Titration Lab Report Example Analysis Section make sure you use your own data! 1a. The strongest acid was acid B with an average concentration of 0.083M. Both acid A and C had lower average concentrations, with acid A having an average concentration of 0.050M and acid C having an average concentration of 0.028 M. 1b. The pH of acid B, using the average concentration of 0.083M was: $-\log(0.083\text{M}) = 1.082\dots$

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